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Cyclic hydrazino amidines were converted to the corresponding aminopyrazolyl derivatives. Ring closure between the amino groups of pyrazoline moieties and NH groups of cyclic amidines afforded the following ring systems: 7,8-Dihydroimidazo[1,2-*e*]pyrazolo[1,5-*a*]-1,3,5-triazines, 8,9-dihydro-7*H*-pyrimido[1,2-*e*]pyrazolo[1,5-*a*]-1,3,5-triazines and 7,8,9,10-tetrahydro[1,3]diazepino[1,2-*e*]pyrazolo[1,5-*a*]-1,3,5-triazines.

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An investigation on the synthesis of polyheterocyclic compounds with hydrazine moieties for possible biological activity produced some novel ring systems.

7,8-Dihydroimidazo[1,2-*e*]pyrazolo[1,5-*a*]-1,3,5-triazines were prepared as follows: 2-Hydrazinoimidazoline (2a) (2) which was obtained by hydrazinolysis of the readily available 2-methylthioimidazoline (1a) (3) was reacted with α -iminobutyronitrile (3a), α -cyanoacetophenone (3b) or α -cyano *p*-methylacetophenone (3c) to give 3'-methyl-2-[5'-aminopyrazolyl]imidazoline (4a) and its 3'-phenyl (4b) and 3'-*p*-tolyl derivatives (4c), respectively. The latter compounds were reacted with *ortho* esters to give substituted 7,8-dihydroimidazo[1,2-*e*]pyrazolo[1,5-*a*]-1,3,5-triazines (7a) (1). Starting with 2-methylthio-1,4,5,6-tetrahydropyrimidine (1b) (4) and 2-methylthio-4,5,6,7-tetrahydro-1*H*-[1,3]diazepine (1c) (4), after hydrazinolysis (2) and interaction of the appropriate 2-hydrazino derivatives, 2b and 2c, with 3a, 3b and 3c the desired 2-[3'-substituted-5-aminopyrazolyl]-1,4,5,6-tetrahydropyrimidines (5a-c) and 2-[3'-substituted-5'-aminopyrazolyl]-4,5,6,7-tetrahydro-1*H*-[1,3]diazepines (6a-c) were obtained, respectively. Substituted aminopyrazolyls 5a-c and 6a-c were reacted with *ortho* esters to give the corresponding 8,9-dihydro-7*H*-pyrimido[1,2-*e*]pyrazolo[1,5-*a*]-1,3,5-triazines (8a-k) and

Scheme 1

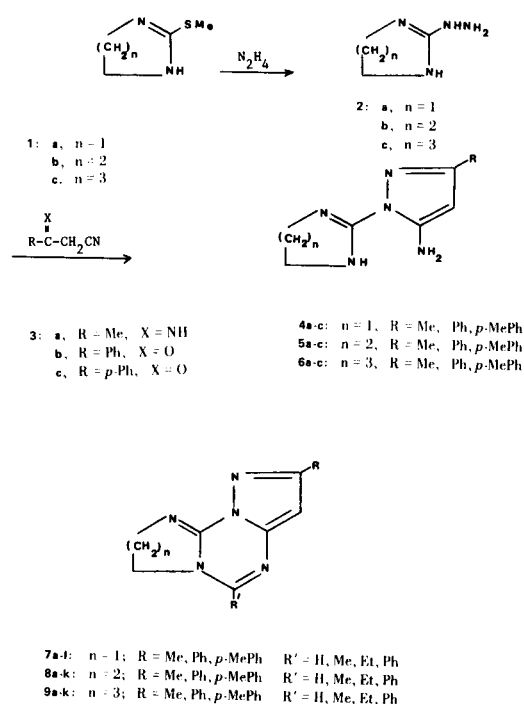
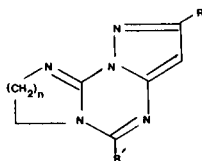


Table I

Compound	n	R	M.p. °C (a)	Yield %	Formula	Analyses					
						C%		H%		N%	
					Calcd.	Found	Calcd.	Found	Calcd.	Found	
4a	1	Me	110	85	C ₇ H ₁₁ N ₅	50.90	50.82	6.66	6.39	42.42	42.49
4b	1	Ph	125	95	C ₁₂ H ₁₃ N ₅	63.43	63.40	5.85	5.66	30.83	31.03
4c	1	<i>p</i> -MePh	105	80	C ₁₃ H ₁₅ N ₅	64.73	64.81	6.22	6.32	29.04	29.11
5a	2	Me	55	84	C ₈ H ₁₃ N ₅	53.63	53.70	7.26	7.26	39.10	39.02
5b	2	Ph	110	91	C ₁₃ H ₁₅ N ₅	64.73	64.69	6.22	6.19	29.04	29.11
5c	2	<i>p</i> -MePh	160	92	C ₁₄ H ₁₇ N ₅	65.88	65.93	6.66	6.41	27.45	27.66
6a	3	Me	80	80	C ₉ H ₁₅ N ₅	55.95	56.11	7.77	7.70	36.26	36.19
6b	3	Ph	110	85	C ₁₄ H ₁₇ N ₅	65.88	65.79	6.66	6.60	27.45	27.33
6c	3	<i>p</i> -MePh	155	75	C ₁₅ H ₁₉ N ₅	66.91	67.06	7.06	7.16	26.02	25.18

(a) Free base.

Table II



Compound	n	R	R'	M.p. °C	Yield %	Formula	Analyses					
							C%		H%		N%	
						Calcd.	Found	Calcd.	Found	Calcd.	Found	
7a	1	Me	H	150	80	C ₈ H ₉ N ₅	54.85	54.91	5.14	5.17	40.00	39.86
7b	1	Me	Me	170-175	85	C ₉ H ₁₁ N ₅	57.14	57.09	5.82	5.96	37.03	37.11
7c	1	Me	Et	175	86	C ₁₀ H ₁₃ N ₅	59.11	58.98	6.40	6.33	34.48	34.44
7d	1	Me	Ph	245	50	C ₁₄ H ₁₃ N ₅	66.93	66.69	5.18	5.19	27.89	27.71
7e	1	Ph	H	236-239	95	C ₁₃ H ₁₁ N ₅	65.82	65.85	4.64	4.71	29.53	29.69
7f	1	Ph	Me	285	79	C ₁₄ H ₁₃ N ₅	66.93	66.83	5.17	5.21	27.88	28.03
7g	1	Ph	Et	218	90	C ₁₅ H ₁₅ N ₅	67.92	67.97	5.66	5.63	26.42	26.55
7h	1	Ph	Ph	212-215	61	C ₁₉ H ₅ N ₅	72.84	72.79	4.79	4.86	22.39	22.22
7i	1	<i>p</i> -MePh	H	265-270	69	C ₁₄ H ₁₃ N ₅	66.93	66.88	5.17	5.09	27.88	27.59
7j	1	<i>p</i> -MePh	Me	290	74	C ₁₅ H ₁₅ N ₅	67.92	67.90	5.66	5.80	26.42	26.46
7k	1	<i>p</i> -MePh	Et	240	77	C ₁₆ H ₁₇ N ₅	68.82	68.69	6.09	5.99	25.09	25.11
7l	1	<i>p</i> -MePh	Ph	205	48	C ₂₀ H ₁₇ N ₅	73.39	73.33	5.20	5.17	21.41	21.41
8a	2	Me	H	120	73	C ₉ H ₁₁ N ₅	57.14	57.18	5.82	5.86	37.03	37.00
8b	2	Me	Me	175	85	C ₁₀ H ₁₃ N ₅	59.11	59.20	6.40	6.33	34.48	34.39
8c	2	Me	Et	195	90	C ₁₁ H ₁₅ N ₅	60.82	60.88	6.91	6.89	32.25	32.32
8d	2	Ph	H	260-262	83	C ₁₄ H ₁₃ N ₅	66.93	67.01	5.17	5.19	27.88	27.91
8e	2	Ph	Me	268	81	C ₁₅ H ₁₅ N ₅	67.92	67.99	5.66	5.60	26.41	26.39
8f	2	Ph	Et	216	87	C ₁₆ H ₁₇ N ₅	68.81	68.80	6.09	6.11	25.08	25.17
8g	2	Ph	Ph	215	49	C ₂₀ H ₁₇ N ₅	73.39	73.44	5.20	5.17	21.41	21.50
8h	2	<i>p</i> -MePh	H	262	69	C ₁₅ H ₁₅ N ₅	67.92	68.01	5.66	5.69	26.41	26.44
8i	2	<i>p</i> -MePh	Me	240	74	C ₁₆ H ₁₇ N ₅	68.82	68.76	6.09	6.13	28.09	25.13
8j	2	<i>p</i> -MePh	Et	235	93	C ₁₇ H ₁₉ N ₅	69.62	69.57	6.48	6.50	23.89	23.79
8k	2	<i>p</i> -MePh	Ph	234	59	C ₂₁ H ₁₉ N ₅	73.90	73.81	5.57	5.55	20.52	20.66
9a	3	Me	H	230	57	C ₁₀ H ₁₃ N ₅	59.11	59.20	6.40	6.37	34.45	34.40
9b	3	Me	Me	250	69	C ₁₁ H ₁₅ N ₅	60.82	60.91	6.91	6.87	32.25	32.32
9c	3	Me	Et	125	76	C ₁₂ H ₁₇ N ₅	62.33	62.40	7.35	7.34	30.30	30.22
9d	3	Me	Ph	218	64	C ₁₆ H ₁₇ N ₅	68.81	68.80	6.09	6.10	25.08	25.11
9e	3	Ph	H	160	52	C ₁₅ H ₁₇ N ₅	67.92	67.82	5.66	5.70	26.41	26.44
9f	3	Ph	Me	215	87	C ₁₆ H ₁₇ N ₅	68.81	68.89	6.09	5.99	25.08	25.02
9g	3	Ph	Et	190	82	C ₁₇ H ₁₉ N ₅	69.62	69.49	6.48	6.50	23.89	23.83
9h	3	MePh	H	180	51	C ₁₆ H ₁₇ N ₅	68.81	68.68	6.09	6.13	25.08	25.06
9i	3	<i>p</i> -MePh	Me	225	76	C ₁₇ H ₁₉ N ₅	69.62	69.69	6.48	6.50	23.89	23.77
9j	3	<i>p</i> -MePh	Et	175	78	C ₁₈ H ₂₁ N ₅	70.35	70.40	6.84	6.90	22.80	22.71
9k	3	<i>p</i> -MePh	Ph	250	55	C ₂₂ H ₂₁ N ₅	74.36	74.39	5.91	5.99	19.71	19.66

7,8,9,10-tetrahydro[1,3]-diazepino[1,2-*e*]pyrazolo[1,5-*a*]-1,3,5-triazines (**9a-k**). (See Scheme I).

The structure of compounds **4-9** were confirmed by elemental analysis, ir, nmr and mass spectroscopy. Compounds prepared are reported in Tables I and II.

EXPERIMENTAL

Melting points were taken on a Kofler hot stage microscope and are uncorrected. The ir spectra were obtained on a Perkin-Elmer 167 spectrophotometer. Nmr spectra were recorded using a Varian T-60A spectrometer and chemical shifts (δ) are in ppm relative to internal tetramethylsilane. Mass spectra were run on a Varian 311 spectrometer at 70 ev.

3'-methyl-2-[5'-aminopyrazolyl]imidazoline (**4a**).

To a solution of 4.5 g. (0.02 mole) of 2-hydrazinoimidazoline hydroiodide (**2a**) in 10 ml. of acetic acid 1.64 g. (0.02 mole) of freshly prepared α -iminobutyronitrile was added and the mixture heated with stirring on a water bath for one half hour. After cooling the yellow crystalline mass was filtered and recrystallized from ethanol to give 5.60 g. (85%) **4a** as hydroiodide, m.p. 220°. The free base was obtained by alkalization of a solution of hydroiodide with dilute sodium hydroxide solution and extraction with chloroform. After evaporation of the solvent, the residue was recrystallized from aqueous ethanol, m.p. 110°; nmr (deuteriochloroform): δ 2.03 (s, 3H, CH₃), 3.62 (s, 4H, CH₂), 5.0 (s, 1H, CH), 5.74 (b, 3H, NH and NH₂); molecular weight by mass spectroscopy, m/e 165.

Compound **4b-c**, **5a-c** and **6a-c** were prepared in a similar man-

ner and are summarized in Table I.

2-Methyl-7,8-dihydroimidazo[1,2-*e*]pyrazolo[1,5-*a*]-1,3,5-triazine (**7a**).

A mixture of 1.65 g. (0.01 mole) of 3-methyl-2-[5'-aminopyrazolyl]imidazoline (**4a**) and 6.72 g. (0.03 mole) of triethylortho-benzoate in 15 ml. of absolute ethanol was refluxed for four hours. After evaporation of the solvent, the solid was recrystallized from ethyl acetate to give 1.25 g. (50%) of **7a**, m.p. 245°; molecular weight by mass spectroscopy, *m/e* 251; uv (methanol): 238 nm (ϵ 4.75), 304 (ϵ 4.32); ir (potassium bromide) 1680, 1578, 1538, 1426, 1368, 1329, 1300, 1198, 1001, 982, 776, 705 and 671 cm^{-1} ; nmr (deutriochloroform): δ 2.33 (s, 3H, CH₃), 4.10 (s, 4H, CH₂), 6.13 (s, 1H, CH), 7.33-7.82 (m, 5H, C₆H₅).

All other compounds of this series and compounds **8a-k** and **9a-k** were prepared similarly except that the refluxing time in the case

of triethylorthoformate, triethylorthoacetate and triethylorthopropionate was only one hour (see Table II).

Acknowledgments.

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REFERENCES AND NOTES

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